

Available online at www.sciencedirect.com



Tetrahedron Letters

Tetrahedron Letters 48 (2007) 941–944

Reaction of functionalized azomethine ylides with acetylenic dipolarophiles: the facile synthesis of functionalized 2*H*- and 1*H*-pyrroles

Keisuke Kawashima,^a Masanori Hiromoto,^a Kyohei Hayashi,^a Akikazu Kakehi,^b Motoo Shiro^c and Michihiko Noguchi^{a,*}

^aDepartment of Applied Molecular Bioscience, Graduate School of Medicine, Yamaguchi University, Tokiwadai, Ube 755-8611, Japan ^bDepartment of Chemistry and Material Engineering, Faculty of Engineering, Shinshu University, Wakasato, Nagano 380-8553, Japan ^cRigaku Corporation, 3-9-12 Matsubaracho, Akishima, Tokyo 196-8666, Japan

> Received 11 October 2006; revised 5 December 2006; accepted 7 December 2006 Available online 26 December 2006

Abstract—The reaction of functionalized azomethine ylides as C-unsubstituted nitrile ylide equivalents with acetylenic dipolarophiles is mentioned. Therein, the initially formed cycloadducts, 2,5-dihydropyrroles, by the reaction of the azomethine ylides with substituted acetylenes, undergo a fission reaction to afford 2*H*-pyrroles and the parent heterocyclic system. Some 2*H*-pyrroles isomerized to 1*H*-pyrroles under both thermal and acidic conditions. © 2006 Published by Elsevier Ltd.

The cycloaddition reaction of N-unsubstituted (NH) azomethine ylides with olefinic dipolarophiles has been a very attractive approach to the N-unsubstituted (NH) pyrrolidines, which are found widely in biologically and pharmacologically active compounds.¹ Typical methods for the generation of NH-azomethine ylides constitute of group rearrangement of α -metalloamides² and 1,2-prototropic³ and N-metalation⁴ routes of the imines of α -amino esters. The former two routes are uncatalyzed thermal processes, however those have not always accomplished the stereoselective cycloaddition of the resulting NH-azomethine ylides because of the harsh generation conditions. In a recent paper,⁵ we reported that the NH-azomethine ylide A, which was generated by the 1,2-prototropic isomerization of the corresponding imine under extremely mild conditions underwent a cycloaddition reaction with N-phenylmaleimide (NPMI) to give a cycloadduct, proline derivative B. stereoselectively. More interestingly, the treatment of the cycloadduct **B** with acetic acid caused a fission reaction to afford dehydroproline derivative C and the parent heterocyclic system D in good yields, the former of which corresponds to the cycloadduct of C-unsubstituted nitrile ylide with NPMI. We, therefore, thought

0040-4039/\$ - see front matter @ 2006 Published by Elsevier Ltd. doi:10.1016/j.tetlet.2006.12.018

that we could propose an equivalent process to C-unsubstituted nitrile ylide cycloaddition reaction (Scheme 1).

Our next concern was directed toward the reaction of the functionalized NH-azomethine vlides with acetylenic dipolarophiles, because only few approaches to the general and effective preparation of 2*H*-pyrroles⁶ had been reported compared with 1*H*-pyrroles. When the solution of aldehyde $\hat{1}$ with (DL)-phenylalanine methyl ester 2a(1.5 equiv) in toluene was heated at 85 °C for 1 h and cooled to room temperature, dimethyl acetylenedicarboxylate (DMAD) was added to the reaction mixture and the mixture was allowed to stir at room temperature for 14 h. After the purification with silica gel column chromatography, the desired 2,5-dihydropyrrole 4a (53%), Michael type adducts 5a (44% as an E/Z mixture), 4-hydroxy-1H-pyrrole 6 (11%) and parent heterocycle 7 (9%) were obtained. Undesired products 5a and 6 were formed from only the reaction of DMAD with amino ester 2a.⁷ The reaction of 1, 2a, and dibenzoylacetylene (DBZA) afforded the almost same results: 2,5-dihydropyrrole 8a (36%), 9a (47%; as an E/Z mixture), and 7 (6%) were formed. Single crystal X-ray structure analysis⁸ for 4a and 8a showed that these were the cycloadducts of DMAD and DBZA to the (E,E)azomethine vlide (E,E)-3a, respectively. Although the

^{*} Corresponding author. Tel.: +81 836 85 9261; fax: +81 836 85 9201; e-mail: noguchi@yamaguchi-u.ac.jp



Scheme 1. Functionalized azomethine ylide A as C-unsubstituted nitrile ylide equivalent.

reaction conditions including solvent, temperature, reagent molar ratio, addition sequences and dehydrating reagent were examined, the yield of the cycloadducts 4a and 8a could not be improved. Gentle heating of 4a and **8a** in toluene gave the desired 2*H*-pyrroles $10a^9$ and 11a in moderate to good yields together with 7. These mean that the reaction sequences are a novel and facile synthetic method for functionalized 2H-pyrroles in mild conditions (Scheme 2). Probably DMAD and DBZA are too reactive toward the amine nucleophile 2a leading to 5a and 9a to accomplish efficiently the imine formation between aldehydes 1 and 2. So, we chose methyl propiolate (MP) as the next acetylenic dipolarophile with less reactivity. The solution of 1 and 2a in toluene was heated at 85 °C for 1 h, cooled down to room temperature, and MP was added to the solution. The reaction mixture was stirred at 15 °C for 3 d.

The reaction underwent smoothly and regioselectively to give cycloadduct **12a** (79%), 1*H*-pyrrole **13a** (9%) and the parent heterocycle 7 (9%) with a depression of the formation of Michael type adduct such as 5a and 9a. The structure of cycloadduct 12a was also confirmed by its single crystal X-ray analysis,8 while that of 1H-pyrrole 13a was determined by the comparison of the spectroscopic data reported in the literature.¹⁰ Heating the isolated **12a** in toluene at 85 °C for 12 h gave 1*H*pyrrole 13a (78%) together with 7 (92%) and a small amount of unidentified product. Standing the chloroform-d solution of 12a at room temperature for 12 days gave a mixture of 2H-pyrrole **14a**¹¹ and **7**. The 2H-pyrrole 14a was not so stable and converted gradually to 1H-pyrrole 13a together with a small amount of an unidentified product on further standing (for 50 days) at the same temperature (Scheme 3).



Scheme 2. The reaction of functionalized azomethine ylide 3a with DMAD and DBZA and thermal behavior of cycloadducts 4a and 8a.



Scheme 3. The reaction of functionalized azomethine ylide 3a with MP.

So, we applied the one-pot procedure to this reaction; the reaction of 1, 2a, and MP in toluene gave 13a and 7 in 71% and 86% yield, respectively (Table 1; entry 1).¹² Stimulated by these findings, the similar reaction of 1, amino esters 2b–f and MP was examined. Although pyridinium *p*-toluene sulfonate (PPTS) was required for completing the reaction in many cases, the desired 1*H*-pyrroles 13b–f and 7 were formed in moderate to good yields. These results are summarized in Table 1.

In order to obtain further information on the reaction path from 12 to 13, the similar one-pot reaction of 1, ethyl propiolate (EP) and 2a, 2b, and 2e was also examined to give 1H-pyrrole 15a, 15b, and 15e in good to

excellent yields as single isomers together with 7 (Table 1; entries 12–14). Fortunately, recrystallization of **15a** from benzene afforded good single crystals for the X-ray structure analysis⁸ and the structure of **15a** was confirmed to be 4-ethyl 3-methyl 2-benzylpyrrole-3,4-dicarboxylate. Although details are still unclear, a plausible pathway from **12** to **13** is demonstrated in Scheme 4; the elimination of the parent heterocycle 7 from cycloadduct **12** takes place thermally to afford 2*H*-pyrroles **14**, which undergo a 1,5-ester group rearrangement accompanied with the aromatization to 1*H*-pyrrole **13**.¹⁵

Our final concern in this one-pot reaction was focused on the generality of acetylenic dipolarophiles; the reac-

Table 1. The reaction of functionalized azomethine ylides 3 with MP and EP in a one-pot method

| | $\begin{array}{c} 1 \\ H_2 N \end{array}$ | 2 CO ₂ Me (1.5-1.8 ec ne, 85 °C, 1 h | uiv.) 3) additive | $Z^{4} = CO_{2}Me^{-4} T (57-98\%)$ | |
|-------|--|--|---|-------------------------------------|-----------------------|
| | 2) HC≡C−Z MP or EP (1.3-1.5 equiv.) 85 °C, reaction time | | 13: Z = CO ₂ Me 15 : Z = CO ₂ Et | | |
| Entry | Amino ester/R | Propiolate/Z | Additive (equiv) | Reaction time (h) | 1H-Pyrroles/yield (%) |
| 1 | Bn (2a) | CO ₂ Me | None | 22 | 13a /71 |
| 2 | . , | - | PPTS (1.0) | 12 | Many products |
| 3 | Ph (2b) | CO ₂ Me | PPTS (0.5) | 2 | 13b /80 |
| 4 | Me (2c) | CO ₂ Me | None | 57 | 13c /40 |
| 5 | | | PPTS (0.5) | 18 | 13c /49 |
| 6 | sec-Bu (2d) | CO ₂ Me | None | 48 | 13d /57 |
| 7 | | | PPTS (0.5) | 6.5 | 13d /87 |
| 8 | <i>i</i> -Bu (2e) | CO_2Me | None | 66 | 13e /27 |
| 9 | | | PPTS (0.5) | 5 | 13e /80 |
| 10 | $(CH_2)_2SMe(2f)$ | CO_2Me | None | 97 | 13f /70 |
| 11 | | | PPTS (0.5) | 45 | 13f /44 |
| 12 | Bn (2a) | CO ₂ Et | None | 20 | 15a /83 |
| 13 | Ph (2b) | CO ₂ Et | PPTS (0.5) | 3.5 | 15b /quant. |
| 14 | <i>i</i> -Bu (2e) | CO ₂ Et | PPTS (0.5) | 2 | 1 5e /93 |

1*H*-Pyrroles **13a**, ¹⁰ **13b**, ¹³ and **13c**¹⁴ are known compounds.



Scheme 4. A plausible pathway from cycloadduct 12 to 1*H*-pyrrole 13.



Scheme 5. The reaction of functionalized azomethine ylide 3a with EPP and PA.

tion of 1, 2a, and ethyl phenylpropiolate (EPP) gave an almost 2:1 mixture of two regioisomeric 2*H*-pyrroles 17a and 18a. Interestingly, the reaction of 1, 2a, and phenylacetylene (PA) gave the desired 1*H*-pyrrole 19a in a moderate yield. Diphenylacetylene (DPA) did not afford any adducts as expected (Scheme 5).

In conclusion, we have reported an effective and versatile preparation method for functionalized 2H- and 1H-pyrroles, in which the cycloaddition reaction of functionalized azomethine ylides as C-unsubstituted nitrile ylide equivalents with acetylenic dipolarophiles leading to cycloadducts, 2,5-dihydropyrroles, is a key step. The resulting 2,5-dihydropyrroles undergo a fission reaction to give 2H-pyrroles and the parent heterocyclic system. The 2H-pyrroles unsubstituted at the 3-position undergo a facile isomerization to 1H-pyrroles. The onepot procedure for functionalized 1H-pyrroles is also accomplished. Further investigation on this chemistry is now underway in our laboratory.

References and notes

- (a) Obst, U.; Betschmann, P.; Lemer, C.; Seiler, P.; Diederich, F.; Gramilich, V.; Weber, L.; Banner, D. W.; Schönholzer, P. *Helv. Chim. Acta* 2000, *83*, 855–909; (b) Pearson, W. H. In *Studies in Natural Products Chemistry*; Atta-Ur-Rahman, Ed.; Elsevier: New York, 1998; Vol. 1, 323–358; (c) Gribble, G. W. In *Comprehensive Heterocyclic Chemistry*; Katritzky, A. R., Rees, C. W., Scriven, E. F. V., Eds.; Pergamon: Oxford, UK, 1996; Vol. 2, 207–257.
- (a) Komatsu, M.; Kasano, Y.; Yonemori, J.-I.; Oderaotoshi, Y.; Minakata, S. *Chem. Commun.* 2006, 526–528; (b) Komatsu, M.; Okada, H.; Yokoi, S.; Minakata, S. *Tetrahedron Lett.* 2003, 44, 1603–1606; (c) Okada, H.; Akaki, T.; Oderaotoshi, Y.; Minakata, S.; Komatsu, M. *Tetrahedron* 2003, 59, 197–205; (d) Komatsu, M.; Okada, H.; Akaki, T.; Oderaotoshi, Y.; Minakata, S. *Org. Lett.* 2002, 4, 3505–3508. Also see references cited therein.
- (a) Garner, P.; Kaniskan, H. U. J. Org. Chem. 2005, 70, 10868–10871; For reviews: (b) Grigg, R. Chem. Soc. Rev. 1987, 16, 89–121; Grigg, R. In New Aspects of Organic Chemistry I; Yoshida, Z., Shiba, T., Oshiro, Y., Eds.; VCH, 1989; pp 113–134; Grigg, R.; Kennewll, P.; Savic, V.; Sridharan, V. Tetrahedron 1992, 48, 10423–10430; (c) Grigg, R.; Gunaratne, H. Q. N.; Kemp, J. Tetrahedron 1990, 46, 6467–6482; (d) Grigg, R.; Donegan, G.; Gunaratne, H. Q. N. Tetrahedron 1989, 45, 1723–1746; For a review: (e) Tsuge, O.; Kanemasa, S. Adv. Heterocycl. Chem. 1989, 45, 232–249.
- A recent review: (a) Husinec, S.; Savic, V. *Tetrahedron: Asymmetry* 2005, *16*, 2047–2061; (b) Gao, W.; Zhang, X.; Raghunath, M. *Org. Lett.* 2005, *7*, 4241–4244; (c) Garner, P.; Kaniskan, H. U. *Tetrahedron Lett.* 2005, *46*, 5181– 5184; (d) Oderaotoshi, Y.; Cheng, W.; Fujitomi, S.; Kasano, Y.; Minakata, S.; Komatsu, M. *Org. Lett.* 2003, *5*, 5043–5046. Also see references cited therein.
- 5. Kawashima, K.; Kakehi, A.; Noguchi, M. *Tetrahedron*, in press.
- For reviews: Sammes, M. P.; Katrizky, A. R. Adv. Heterocycl. Chem. 1982, 32, 233–284; Patterson, J. M. Synthesis 1976, 281–304.
- 7. Kolar, P.; Tišler, M. Synth. Commun. 1994, 24, 1887-1893.
- 8. Structures of 2,5-dihydropyrroles **4a**, **8a**, and **12a** and 1*H*-pyrrole **15a** were confirmed by single crystal X-ray

structure analysis and their crystallographic data have been deposited with the Cambridge Crystallographic Data Centre as supplementary publication numbers CCDC 621689 for **4a**, CCDC 621692 for **8a**, CCDC 621691 for **12a**, and CDDC 621693 for **15a** copies of the data can be obtained, free of charge, on application to CCDC, 12 Union Road, Cambridge CB2 1EZ, UK (fax: +44 1223 336033 or e-mail: deposit@ccdc.cam.ac.uk).

- The solution of 4a (0.350 g, 0.672 mmol) in toluene (5 mL) was heated for 20 h and the solvent was evaporated to dryness, which was subjected to silica-gel column chromatography to afford 10a (0.180 g, 81%) as an eluent of hexane/ethyl acetate (3:1) and 7 (0.126 g, quant.) as an eluent of hexane/ethyl acetate (1:2), respectively. 2*H*-Pyrrole 10a: Colorless oil; ¹H NMR (CDCl₃): 3.50, 3.79 (each 1H, each d, *J* = 13.5 Hz, CH₂Ph), 3.64, 3.71, 3.86 (each 3H, each s, 3 × OCH₃), 7.03–7.11 (5H, m, Ph–H), 8.02 (1H, s, 5-H); ¹³C NMR (CDCl₃): 40.25 (CH₂Ph), 52.49, 52.81, 53.32 (3 × OCH₃), 92.06 (2-C), 127.10, 127.60, 130.35, 133.15, 135.71, 159.17 (Ph–C and 3- and 4-C), 160.99, 163.02, 163.38 (3 × CO₂CH₃), 166.81 (5-C).
- Padwa, A.; Gasdaska, J. R.; Hoffmanns, G.; Rebello, H. J. Org. Chem. 1987, 52, 1027–1035.
- 11. 2*H*-Pyrrole **14a**: ¹H NMR (CDCl₃): 3.37, 3.54 (each 1H, each d, J = 13.5 Hz, CH_2 Ph), 3.68, 3.81 (each 3H, each s, $2 \times OCH_3$), 7.11–7.36 (5H, m, Ph–H), 8.04, 8.27 (each 1H, each d, J = 0.7 Hz, 5- and 3-H); ¹³C NMR (CDCl₃): 41.44 (*C*H₂Ph), 52.11, 53.07 (2 × OCH₃), 91.27 (2-C), 127.24, 128.03, 128.93, 130.03, 134.29, 135.22 (Ph–C and 3- and 4-C), 160.96, 164.04 (2 × CO₂CH₃), 168.01 (5-C). Chromatographic separation on silica-gel of **14a** from the mixture gave an inseparable mixture of **14a**, **13a**, and other many products due to the decomposition.
- 12. The solution of 1 (0.143 g, 0.658 mmol) and 2a (0.213 g, 1.19 mmol) in toluene (1 mL) was heated at 85 °C for 1 h. MP was added to the solution, then the mixture was heated at 85 °C for additional 22 h and the reaction mixture was evaporated to dryness. The residue was subjected to silica-gel column chromatography to afford 13a (0.128 g, 71%) as an eluent of hexane/ethyl acetate (3:1) and 7 (0.107 g, 86%) as an eluent of hexane/ethyl acetate (1:2), respectively. 1H-Pyrrole 13a: Colorless crystals from hexane-benzene; mp 156-157 °C (lit.10 142-143 °C); ¹H NMR (CDCl₃): 3.79, 3.84 (each 3H, each s, 2×OCH₃), 4.21 (2H, s, CH₂Ph), 7.14 (1H, d, J = 3.0 Hz), 5-H), 7.19–7.24 (5H, m, Ph–H), 8.26 (1H, br, NH); ¹³C NMR (CDCl₃): 32.99, 51.52, 51.55, 112.31, 116.40, 123.02, 127.06, 128.96, 129.00, 137.32, 137.48, 164.49, 165.33.
- (a) Washikazu, K.-I.; Minakata, S.; Ryu, I.; Komatsu, M. *Tetrahedron* 1999, 55, 12969–12979; (b) Tsuge, O.; Kanemasa, S.; Matsuda, K. J. Org. Chem. 1986, 51, 1997– 2004; (c) Tsuge, O.; Kanemasa, S.; Matsuda, K. Chem. *Lett.* 1985, 1411–1414. Also see, Refs. 2a and 10.
- (a) Padwa, A.; Gasdaska, J. R.; Tomas, M.; Turro, N. J.; Cha, Y.; Gould, I. R. J. Am. Chem. Soc. **1986**, 108, 6739– 6746; (b) Gabel, N. W. J. Org. Chem. **1962**, 27, 301–303.
- 15. A similar 1,5-ester group rearrangement was proposed in the course of isomerization of 2*H*-pyrrole to 1*H*-pyrrole derivatives.^{3c} Therein, the 2*H*-pyrroles were formed in situ by the DDQ-oxidation of 2,5-dihydropyrroles, which were obtained by the reaction of NH-azomethine ylides with MP. Also, an isomerization process, a sigmatropic rearrangement, of 3*H*-pyrrole to 2*H*- and 1*H*-pyrrole finally through the ester group imigration was discussed: Chiu, P.-K.; Sammes, M. P. *Tetrahedron* 1990, 46, 3439–3456; Chiu, P.-K.; Sammes, M. P. *Tetrahedron Lett.* 1987, 28, 2775–2778.